Executive Overview of the REHOS Technology

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Abstract:

The REHOS cycle consist of two different sub-cycles, combined regeneratively, namely an Absorption Heat Transformer (AHT), pumping low-grade heat from the environment, up to higher temperature, and a simple Organic Rankine Cycle (ORC) utilizing the pumped, higher temperature heat to power a turbine. The AHT is enhanced by the addition of a heat pump compressor, forming an AHT Heat Pump with a coefficient of performance very much higher than the conventional vapour compression (VC) heat pumps. The ORC use the pumped higher temperature heat to power the turbine, and the turbine low pressure exhaust vapour is added to the AHT reactor bottom, where it is absorbed regeneratively, instead of rejecting the latent heat to an external cooling system. This regeneration provide (or rather, recirculate) a large portion of the heat required by the ORC, and only the energy shortfall (after complete heat balance of the cycle) is required to be added by external sources.

The heat to power conversion efficiency can therefore be calculated by the ORC Output Power (with pumping and compressor power subtracted), divided by this energy shortfall, and may thus be unbelievably high. This is also described on my website homepage in the second and third paragraphs by way of an actual example:

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Introduction:

REHOS Technology, as well as costing methodologies, was already comprehensively described in several documents (see references at the end of this paper), but it is required to give an uncluttered executive summary to easily understand the revolutionary REHOS concepts.

Summary of REHOS concepts:

The heat of solution (HOS) concepts all stem from the physical characteristics of a zeotropic binary mixture in the vertical column reactor. This is sketched in Figure 1 below and the sketch accentuate the binary mixture reaction with heat added. On the left is the column at ambient conditions, with the 50% ammonia in aqua mixture filling the complete length of the column.

Figure 1 **The Binary Column Ambient Condition Energized Condition** 99% NH3 50% NH3 -20C 25C VH3 Concentration Gradient **Temperature Gradient Zeotropic** Zeotropic External **Binary Binary** Liquid Heat Liquid Added Column Column **Ammonia Ammonia** in in Water Water 5% NH3 100C 50% NH3 25C

As soon as external heat is supplied to the column, the mixture react by creating an ammonia concentration gradient with the higher NH3 concentration moving to the top of the reactor while the higher density, hotter, but leaner NH3 mixture migrating to the bottom of the reactor. Simultaneously the mixture migrating to the top of the reactor is cooled (due to the endothermic process of desorption) while the reactor bottom is heated, creating a temperature gradient with the hottest area at the reactor bottom. Heat energy is required from external sources to establish this concentration and temperature gradient, but once established, the external heat source may be removed and the gradients would very slowly dissipate as heat is lost by radiation out from the hot bottom. Note that due to the saturated nature of the binary column, external heat may enter the column anywhere along its length (therefore also at any temperature greater than the column temperature at the heat entry point), with the same effect.

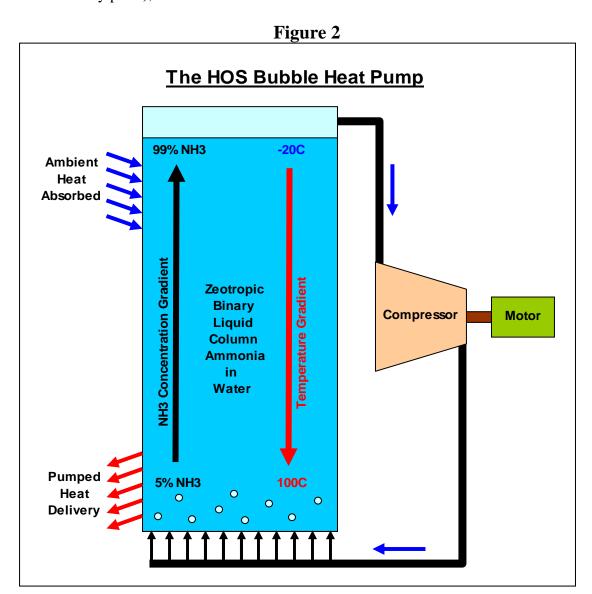
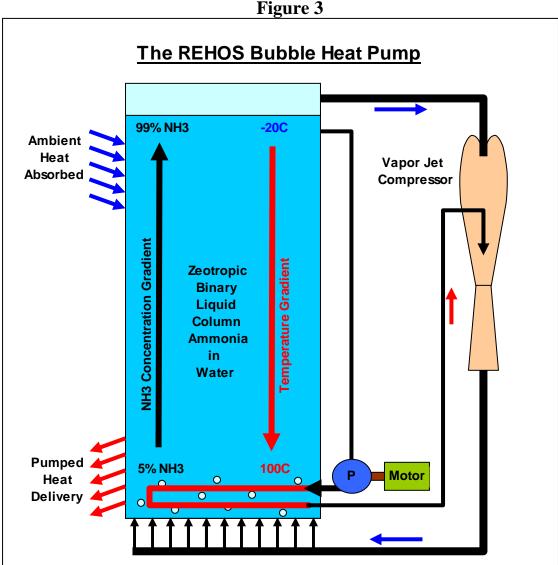


Figure 2, above, sketch the heat of solution (HOS) bubble heat pump. The coefficient of performance (COP) of this type of heat pump is very high, as the compressor pressure

ratio (and therefore temperature increase with pressure), is very small. The compressor only need to provide enough pressure to overcome the liquid column hydraulic pressure created by the gravity force on the liquid column. With the column mixture operating at the temperatures as shown, the compressed vapor temperature only increase isentropically about 2 - 4°C, making the heat pump COP easily go to about 50-100 (still only ~ 80% of the ideal, carnot COP for this small temperature delta). Cold compressed vapor bubbling into the heat transformer reactor is absorbed by the binary mixture, delivering heat to the reactor bottom and raising the reactor bottom temperature by 40 -70°C (typical single stage AHT performance). The exceptionally high COP value (of 50 -100), therefore does not only supply a heat pump operating with the small temperature delta of 2 - 4°C, but through the AHT action in the bubble reactor generate a temperature delta of some 40 - 70°C, which is a factor of at least 20 x the performance of a conventional vapor compression heat pump which would have given a COP ~ 2.9 - 5, for the same $40 - 70^{\circ}$ C temperatures. This give the **HOS heat pump a performance of ~ 20** times the conventional vapor compressor type heat pump.



The regenerative heat of solution (REHOS) bubble type heat pump simply replace the motor driven isentropic compressor with a vapor ejector type compressor, using regenerative heat of absorption of the vapor bubbles at the bottom of the reactor to evaporate the pumped, high pressure liquid in a heat exchanger positioned in the hot bottom of the bubble reactor. This injector type of compressor therefore use a very small amount of electrical energy to only drive the liquid pressure pump. The REHOS type heat pump COP is therefore a factor of several times the HOS heat pump performance, putting it's performance into the range of COP = 350, or a factor of ~ 100 times the conventional vapor compression type heat pump.

The REHOS power cycle sketched in Figure 4 below, differ from the REHOS heat pump only in the removal of a heat output, and use of the pumped heat high temperature energy to evaporate a lot more liquid, to power a vapor expansion turbine to generate power. The low pressure exhaust of the turbine adds to the compressor output vapor, to be regeneratively re-used in the reactor bottom for the evaporation process. The **REHOS** power cycle therefore have a heat to electricity conversion efficiency of 80 - 90%, making it the ideal low temperature waste heat to power energy recovery machines.

Figure 4 The REHOS Power Cycle 99% NH3 -20C **Ambient** Heat Vapor Jet **Absorbed** Compressor Concentration Gradien Zeotropic Electrical **Binary** Power Liquid Output Column **Ammonia** in Water Generator **Power** 100C Motor **Turbine**

Environmental heat can be extracted from the environment, as the cold end of the reactor operate at some 20°C below zero. A simple liquid-liquid heat exchanger can therefore extract ambient heat from water sources like rivers, streams, dams, lakes and the sea to generate power from, at the revolutionary high (> 80%) REHOS cycle efficiency!

Extracting heat from a liquid like water, has a very high power density due to the large medium density and sensible heat content of the ambient temperature water. The heat exchanger to do this is therefore small and cheap, while extraction of heat from the air has a much lower density and heat capacity. The heat exchangers are therefore much more bulky and expensive. In cases where very small amount of power would be required, it would be practical, however.

The usefull side effect of extracting heat from ambient air would obviously deliver chilled air (for air conditioning and refrigeration use) as well as a de-humidifying effect, condensing water vapor from humid ambient air.

References

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- 2. A Paper titled "The Simplified REHOS Cycle.pdf" was written by Johan Enslin in August 2017 and published on my website http://www.heatrecovery.co.za/.cm4all/iproc.php/The Simplified REHOS Cycle.pdf
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